

LESSON PLAN (2024-25)

Name of Assistant Professor : Dr. Satish Kumar

Sem. : 1st

Subject : M.Sc. Chemistry

Weeks	Date	Organic Spectroscopy
1 st	22 July - 27 July	Introduction, ion production - EI, CI, FD and FAB, factors affecting fragmentation
2 nd	29 July- 03 Aug	Ion analysis, Ion abundance, mass spectral fragmentation of organic compounds, Common functional groups
3 rd	05 Aug- 10 Aug	Molecular ion peak, Metastable peak, Nitrogen rule, Molecular weight determination, Molecular formula from isotopic ratio data.
4 th	12 Aug - 17 Aug	Isotope profile of halogen compounds, factors affecting reaction pathways,
5 th	19 Aug- 24 Aug	Fragmentation pattern - simple cleavage, retro Diels Alder, Hydrogen transfer rearrangement like scrambling, ortho effect
6 th	26 Aug- 31 Aug	McLafferty rearrangement, Fragmentation of Hydrocarbons, Alcohols, Phenols, ethers, Aldehydes
7 th	02 Sep - 07 Sep	Ketones, esters, carboxylic acids, amines, nitro, amides, nitriles
8 th	09 Sep - 14 Sep	Revision of Mass spectrometry
9 th	16 sep- 21 sep	Assignment and Unit tests
10 th	24 sep- 28 sep	Carbon 13 nucleus, chemical shifts and its calculation
11 th	30 sep- 05 oct	Proton coupled and decoupled C-13 Spectra. NOE
12 th	07 oct - 12 oct	Cross polarisation, problems with integration
13 th	14 oct -19 oct	Molecular relaxation process, off resonance decoupling
14 th	21 oct - 26 oct	Distortionless enhancement by polarization transfer,
15 th	28 oct -02 nov	Diwali Break
16 th	04 nov- 09 nov	Heteronuclear coupling of CARBON to deuterium, fluorine and phosphorus
17 th	11 nov - 16 nov	IR, NMR, UV and conjoint IR-UV/VIS-NMR - MASS spectrometry problems
18 th	18 nov- 23 nov	IR, NMR, UV and conjoint IR-UV/VIS-NMR - MASS spectrometry problems
19 th	25 nov - 30 nov	Revision

Name of Assistant Professor : Ravi Kant

Sem. : 1st

Subject : Chemistry		
Weeks	Date	Topics
1 st	22 July - 27 July	Stereochemistry-I : Introduction to molecular symmetry and chirality
2 nd	29 July- 03 Aug	D-L, R-S, E-Z and threo-erythro nomenclature
3 rd	05 Aug- 10 Aug	Interconversion of Fischer, Newman and Sawhorse and flying wedge formulae
4 th	12 Aug - 17 Aug	Conformational analysis
5 th	19 Aug- 24 Aug	eneatiomerism and diastereomerism of simple acyclic, cyclic system (Chair and boat configuration)
6 th	26 Aug- 31 Aug	fused and bridged bicyclic systems (decalins) and sugars
7 th	02 Sep - 07 Sep	Conformation and reactivity with examples, optical activity in the absence of chiral carbon (biphenyls)
8 th	09 Sep - 14 Sep	Allenes, ansa compounds, cyclophanes, hemispiranes and spiranes
9 th	16 sep- 21 sep	Stereochemistry of the compounds containing Nitrogen, sulphur and phosphorous
10 th	24 sep- 28 sep	Doubt Class and Unit test
11 th	30 sep- 05 oct	Stereochemistry-II : Topicity of ligands and faces, their nomenclature and prostereoisomerism
12 th	07 oct - 12 oct	Stereogenicity, chirogenicity, pseudoasymmetry and prochiral centre
13 th	14 oct -19 oct	Stereospecific and stereoselective reaction, Asymmetric synthesis
14 th	21 oct - 26 oct	Asymmetric synthesis : Enantiomeric excess, % enantioselectivity, optical purity.
15 th	28 oct -02 nov	Diwali Break
16 th	04 nov- 09 nov	Asymmetric synthesis : % diastereomeric excess and % diastereoselectivity
17 th	11 nov - 16 nov	Asymmetric synthesis: Basic principle, auxiliary, substrate, reagent and catalyst controlled
18 th	18 nov- 23 nov	Submission of Assignments by Students
19 th	25 nov - 30 nov	Revision

Name of Assistant Professor : Dr. Anuj		
Class : M.Sc. Chemistry		
Sem. : 1st		
Subject : Physical Chemistry		
Month	Week	Topic
		Unit III :- Chemical Dynamics
	Week 1	Brief description of integrated rate laws of zero and first order

Sept	Week 2	Brief description of integrated rate laws of second order and graphical representation
	Week 3	Lindemann Hinshewood treatment for unimolecular gas; rate law for opposing reaction
	Week 4	Rate law for opposing reactions; parallel reaction
Oct	Week 1	Rate law for consecutive reactions
	Week 2	Chain reactions
	Week 3	Chain reactions, chain length, apparent activation energy
	Week 4	Diwali Vacation
Nov	Week 1	UNIT-IV Ion - Ion Interactions: The Debye-Huckel theory of ion- ion interactions, ionic cloud, Poisson's equation, excess charge density, Linearization of Boltzmann equation
	Week 2	Linearized Poisson Boltzmann equation and its solution, excess charge density and potential as a function of distance from central ion, Debye Huckel reciprocal length ionic cloud and its contribution to the total potential
		Debye-Huckel limiting law of activity coefficients, its physical significance and its limitations, ion-size effect on potential, ion-size parameter and the theoretical mean - activity coefficient in the case of ionic clouds with finite - sized ions.
	Week 4	Debye - Huckel -Onsager treatment for aqueous and non aqueous solutions and its limitations. Debye-Falkenhagen effect, Wein effect.

LESSON PLAN (2024-25)		
Name of Assistant Professor : Anil kumar		
Class : M.Sc.		
Sem. : 1st		
Subject : Chemistry		
Weeks	Date	Topics
1 st	22 July - 27 July	NMR:- introduction , nuclear spin states , nuclear magnetic moments, resonance, population densities.
2 nd	29 July- 03 Aug	chemical shift and shielding mechanism, chemical equivalence , integral and integration,, chemical environments and chemical shift, local diamagnetic ,shielding.

3 rd	05 Aug- 10 Aug	magnetic anisotropy , spin -spin splitting , pascal triangle , comparision of spectra at low and high fields strength , spin spin coupling,role of magnetic equivalence,
4 th	12 Aug - 17 Aug	concept of non equivalence within a group,measuring coupling constants from first order and second order spectra,coupling in aromatic and heteroaromatic system ,
5 th	19 Aug- 24 Aug	spectra of homotopic ,enantiotopic and diastereotopic system,NMR absorption of organic compounds :- alkanes, alkenes, aromatic compounds,alkynes,alkylhalides, alcohols,
6 th	26 Aug- 31 Aug	ethers amines , nitrile aldehydes , ketones, esters, carboxylic acid, amide, nitroalkanes. Discussion on simplification technique- deterium exchange ,
7 th	02 Sep - 07 Sep	chemical shifts reagents, chiral resolving agents, spin decoupling methods, double resonance and NOE spectra.
8 th	09 Sep - 14 Sep	Revision and Doubt sessions. Unit Test For Assessment
9 th	16 sep- 21 sep	carbon NMR:- carbon 13 nucleus, chemical shift, and its calculation, proton coupled and decoupled car-13, nuclear overhauser , enhancement , cross, polarization,
10 th	24 sep- 28 sep	problem with integration , molecular relaxation process , off resonance , decoupling , distortionless enhancement by polarization transfer,
11 th	30 sep- 05 oct	hetero nuclear coupling of carbon to deuterium , fluorine and P
12 th	07 oct - 12 oct	IR spectra:- principal , units of frequency, wavelength and wavenumber ,
13 th	14 oct -19 oct	molecular vib, factor influencing vibrational freq
14 th	21 oct - 26 oct	instrumentation:- dispersive and interferometric ,
15 th	28 oct -02 nov	Diwali Break
16 th	04 nov- 09 nov	sample technique, application of IR, identity by fingerprinting and functional groups
17 th	11 nov - 16 nov	quantitative IR analysis , attenuated total reflectance and multiple internal reflectance
18 th	18 nov- 23 nov	Submission of Assignments by Students
19 th	25 nov - 30 nov	Revision

LESSON PLAN (2024-25)

Name of Assistant Professor : Usha

Subject : Chemistry

Lecture Time : 09:00 to 10:00AM

Weeks	Date	Topics
1 st	08-Aug	Unit :3 - CFT and its limitations
2 nd	14-Aug	Crystal field effects: d- orbital splitting in Oh, Sq. planar complexes

3 rd	21-22 aug	d- orbital splitting in sq. pyramidal and trigonal bipyramidal complexes
4 th	28-29 aug	JT distortion
5 th	04-05 sept	MOT of sigma bonding
6 th	11-12sept	MOT for pi - bonding
7 th	18-19 sept	MOT for pi- bonding
8 th	25-26 sept	Unit : 2 - Inert and labile complexes
9 th	03-Oct	Kinetic applications of VBT and CFT
10 th	09-10 oct	Class Test ans assignment submission
11 th	16-17oct	<u>kinetics of Oh substitution</u>
12 th	23-24 oct	acid hydrolysis and base hydrolysis
13 th	27oct- 03 nov	<u>Diwali Holidays</u>
14 th	06-07 nov	conjugate base mechanism , anation reaction
15 th	13-14nov	racemization of tris chelate complexes
16 th	20-21 nov	reaction without metal ligand bond cleavage

LESSON PLAN 2nd Semester

Name of Assistant Professor : Dr. Satish Kumar

Subject : Organic Chemistry II

Weeks	Date	Topics
1	13 Jan- 18 Jan	Arenium ion mechanism, orientation and reactivity, Energy profile diagram
2	20 Jan - 25 Jan	The ortho/para ratio, ipso attack, orientation in other ring system, Diazonium coupling
3	27 Jan - 01 Feb	Gattermann Koch Reaction, Vilsmeier Haack reaction, Riemer Tiemann reaction, Fries rearrangement
4	03 Feb - 08 Feb	Aromatic nucleophilic substitution, Benzyne and SRN1 Mechanism, effect of substrate structure
5	10 Feb - 15 Feb	leaving group and attacking nucleophile, Thevon Richter and smiles rearrangement.
6	17 Feb - 22 Feb	E1 and E2 and E1CB mechanism, Orientation effect in elimination effect
7	24 Feb - 28 Feb	Reactivity and effect of substrate structure, attacking base, leaving group and the medium
8	03 March -08 March	Saytzeff and Hoffmann rules, Stereochemistry of E2 reaction. And eclipsing effect in E2 Elimination
9	10 March - 15 March	Holi Break
10	17 March-22 March	Submission of Assignment by students Unit Test for Assessment

11	24 March- 29 March	Mechanistic and stereochemical aspects of addition reaction involving electrophile, Nucleophile and free radical
12	31 March - 05 April	Regio and chemoselectivity orientation and reactivity.
13	07 April -12 April	Hydrogenation of double and triple bonds
14	14 April - 20 April	Hydrogenation of aromatic rings
15	21 April - 27 April	Hydroboration, Michael reaction, sharpless asymmetric epoxidation.

LESSON PLAN

Name of Assistant Professor : Usha

Subject : Chemistry

Lecture Timing : 09:00 to 10:00AM

Weeks	Date	Topics
1st(Unit-2)	15-16 jan	Electronic arrangements of microstates, calculation of number of microstates
2 nd	22-23 jan	spectroscopic term symbol, determination of ground state term symbol- Hunds rule
3 rd	29-30jan	correlation and spin orbital coupling in free ions of 1st transition series
4 th	05-06 feb	Orgel and T-S diagram
5 th	13-Feb	Calculation of CFSC, Racah parameter and nephelaunxetic ratio
6 th	19-20 feb	Structural evidence of electronic spectrum, spectrochemical series and nephelauxetic series
7 th	27-Feb	charge transfer spectra, anomalous magnetic moment
8 th	05-06 march	magnetic exchange coupling and cross over, Unit Test and assignment submission
9 th	09-16 march	Holi Break
10th(4th)	19-20 march	Structure and bonding in metal carbonyl clusters
11 th	26-27 march	corundum, pervoskite, limenite, calcite
12 th	02-03 april	higher boranes , carboranes and metalloboranes
13 th	09-10 april	metalloc arboranes and halide clusters, normal and inverse spinels
14 th	23-24 april	fluorite, antifuorite,rutile, antirutile, cristobalite
15 th	30-Apr	layer lattices: CdI ₂ , BiI ₃ , ReO ₃ , M ₂ O ₃ ,

LESSON PLAN

Name of Assistant Professor : Sandeep

Subject : Chemistry

Sem. : 2

Weeks	Date	Topics
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1 st	20 Jan- 25 Jan	Thermodynamics: Third law of thermodynamics, Nernst heat theorem, concept of absolute entropy.
2 nd	27 Jan - 01 Feb	Heterogeneous equilibrium, Phase rule,
3 rd	03 Feb - 08 Feb	Clausius- Clapeyron Equation,
4 th	10 Feb - 15 Feb	Phase diagram for one component
5 th	17 Feb - 22 Feb	system ($H > 0$ & S), two completely miscible components,
6 th	24 Feb - 28 Feb	two components system with congruent and incongruent melting point,
7 th	03 March -08 March	Calculation of eutectic point
8 th	10 March - 15 March	Holi Break
9 th	17 March-22 March	Submission of Assignment by students Unit Test for Assessment
10 th	24 March- 29 March	Chemical Dynamics: Effect of temperature on reaction rates , Arrhenius equation,
11 th	31 March - 05 April	collision theory of reaction rates
12 th	17 March-22 March	thermodynamic formulation of activated complex theory
13 th	07 April -12 April	correlation between various theories of reaction rate
14 th	14 April - 19 April	Enzymatic reaction: Michaelis - Menton treatment, evaluation of Michaelis's constant for enzyme - substrate
15 th	21 April - 26 April	binding by Lineweaver - Burk plot, Concept of inhibition

LESSON PLAN

Name of Assistant Professor : Ravi Kant

Subject : Organic Chemistry II

Weeks	Date	Topics
1	13 Jan- 18 Jan	Reactivity of carbonyl compounds towards addition. Addition of Grignard reagent
2	20 Jan - 25 Jan	Organozinc and organolithium reagents to carbonyl and unsaturated carbonyl compounds
3	27 Jan - 01 Feb	Wittig reaction, Mechanism of metal hydride reduction of saturated and unsaturated carbonyl compounds
4	03 Feb - 08 Feb	Acids, esters and nitriles. Mechanism of condensation reactions involving enolates like Aldol, Knoevenagel
5	10 Feb - 15 Feb	Claisen, Mannich, Robinson, Reformatsky, Benzoin
6	17 Feb - 22 Feb	Perkin and Stobbe reactions. Hydrolysis of esters and amides

7	24 Feb - 28 Feb	Classification and general mechanistic treatment of nucleophilic, free radical rearrangement
8	03 March -08 March	Classification and general mechanistic treatment of Electrophilic rearrangement. Wagner Meerwein reaction
9	10 March - 15 March	Holi Break
10	17 March-22 March	Submission of Assignment by students Unit Test for Assessment
11	24 March- 29 March	Pinacol Pinacolone, Benzil - benzilic acid,
12	31 March - 05 April	Favorskii, Steven and Wittig reaction
13	07 April -12 April	Neber, Wolf, Hoffmann
14	14 April - 20 April	Curtius, Lossen, Schmidt and Bayer -Villiger Reaction
15	21 April - 27 April	Revision

